

Metal-Directed Multiprotein Structures for Redox Catalysis by Copper

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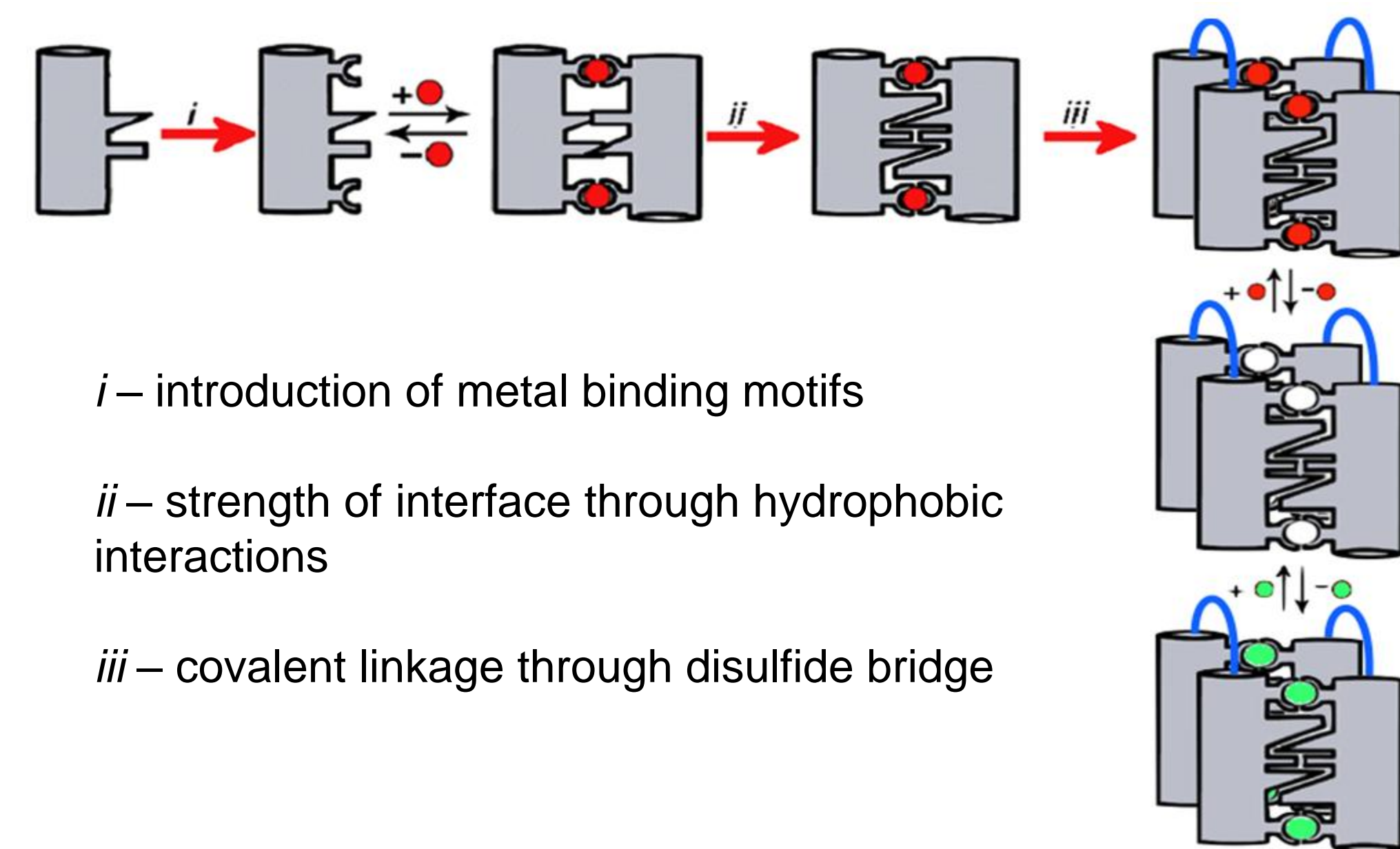
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Abstract

We have demonstrated that we can control protein self-assembly by using metal coordination chemistry. Our goal is to design a copper-binding protein oligomer with tunable redox reactivity. The model protein cytochrome *cb*₅₆₂ was modified by introducing metal binding residues, like histidine and aspartate, which allowed the protein to tetramerize in the presence of metal. Further modifications to this model, by using computational interface redesign, gave rise to a stable metal-free tetramer that we can further use as a template for our studies (known as RIDC-1 96C). We have been making mutations at the tetrameric interface including sulfur-containing amino acids like methionine because these have a high preference for Cu(I) binding. By using analytical ultracentrifugation and protein crystallography we are able to get a clear picture of the copper coordination environment. Once this is understood, we will study if these copper centers are capable of oxygen activation and catalysis.

Overview

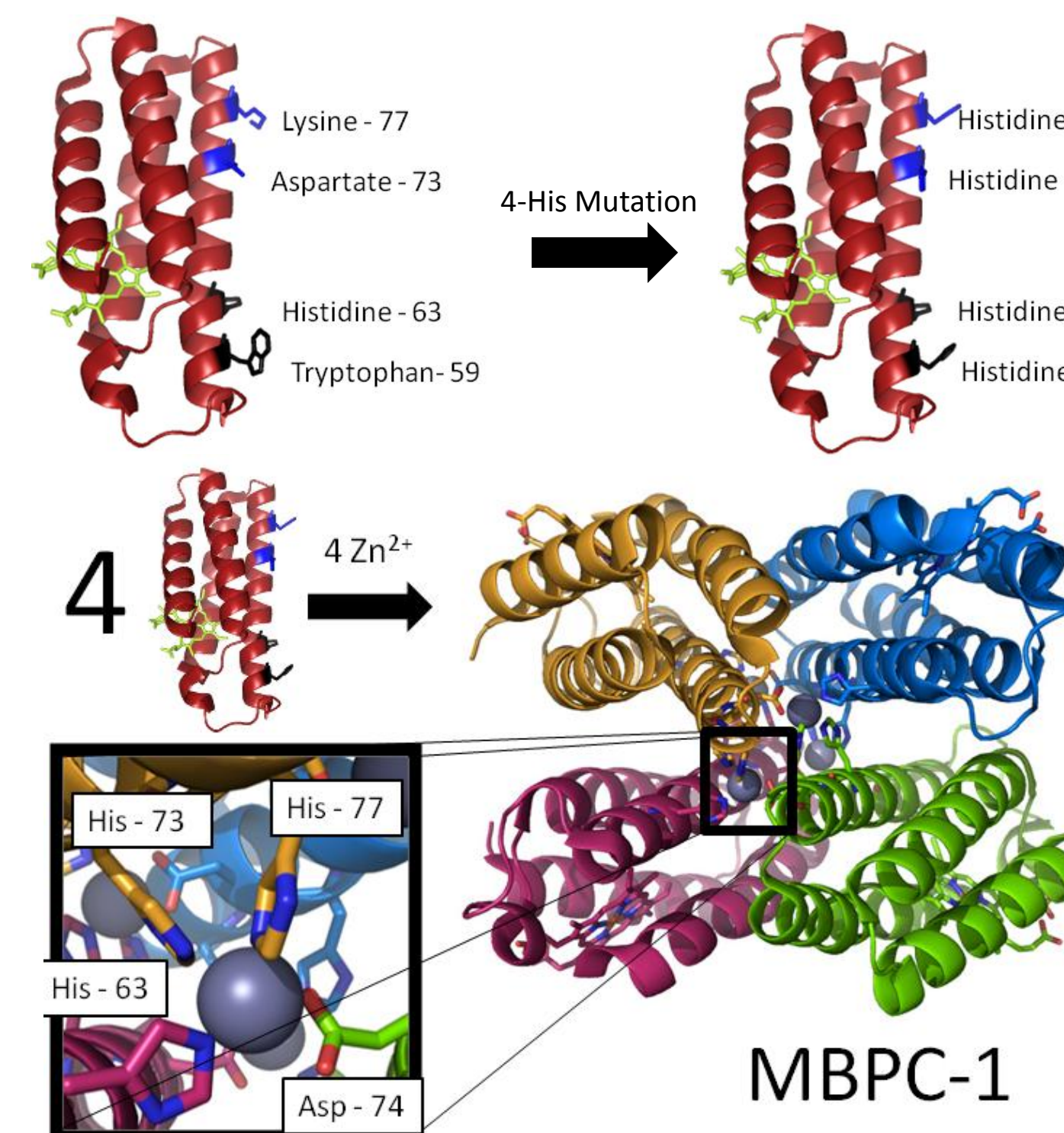


Cytochrome *cb*₅₆₂

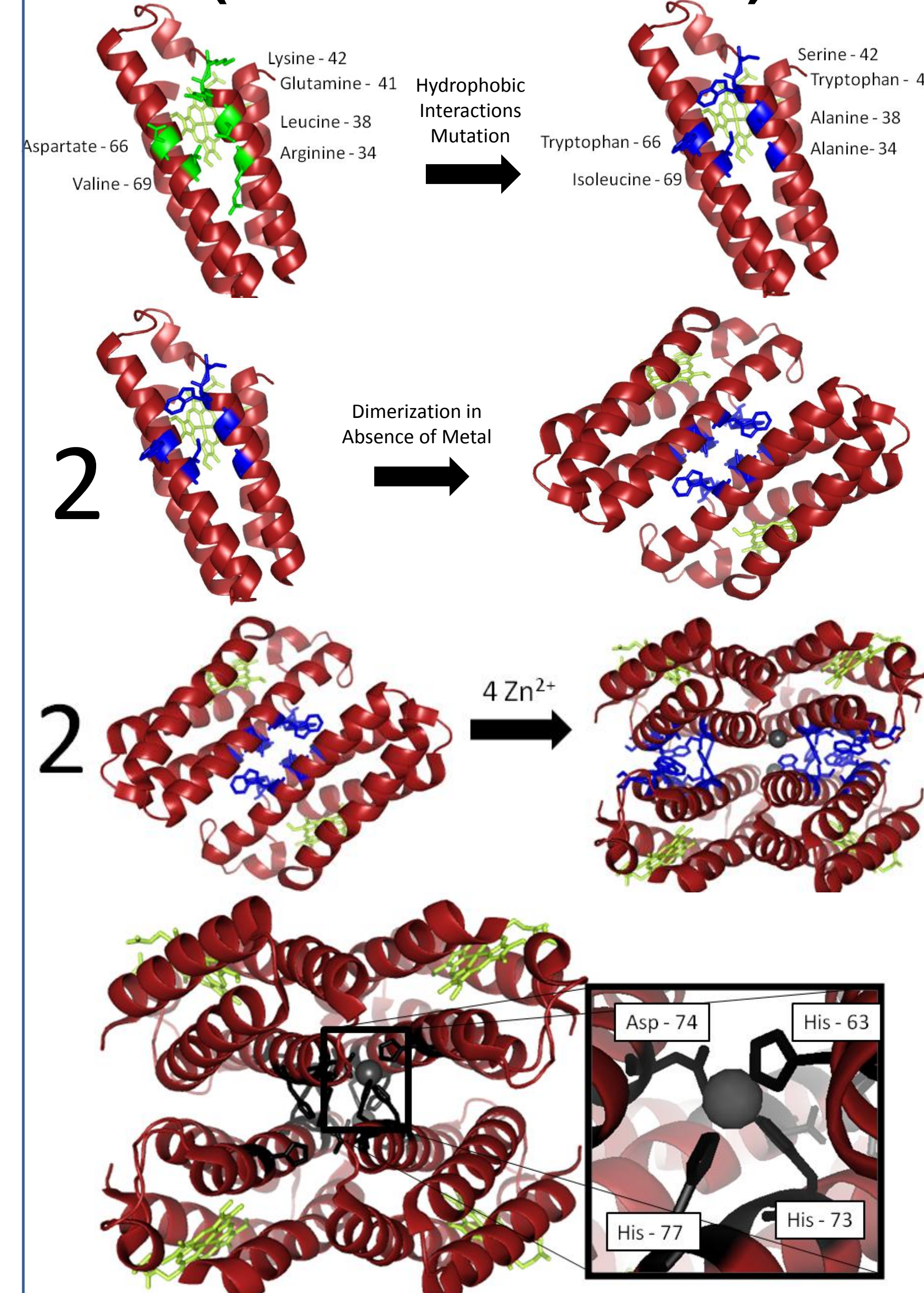


- Four-helix bundle protein
- Stable
 - Temperature
 - pH
- Recombinantly expressed in large quantities
- Monomeric at millimolar concentrations
- Red protein, visible

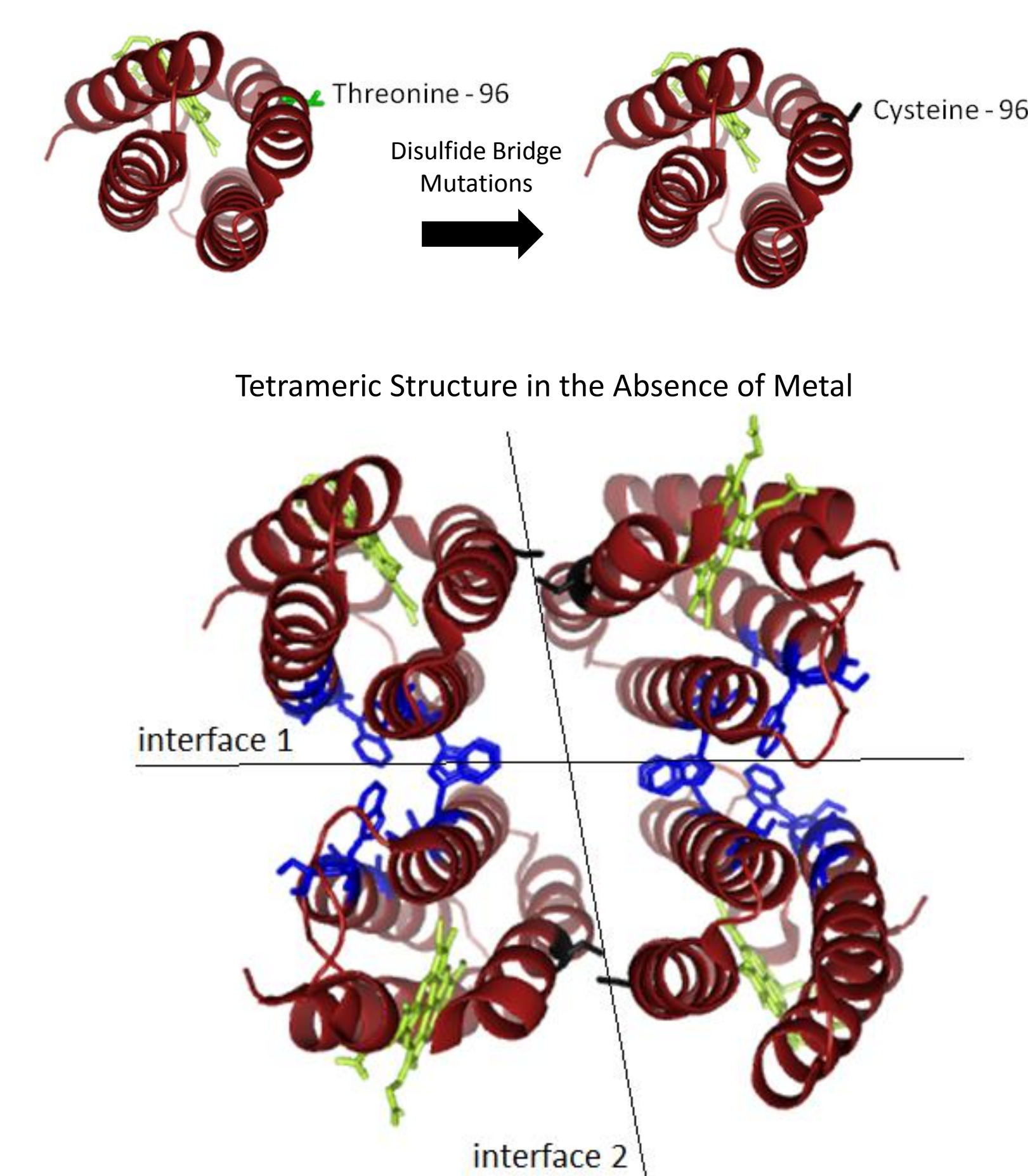
Tetramerization of 4-His Mutation (known as MBPC-1)



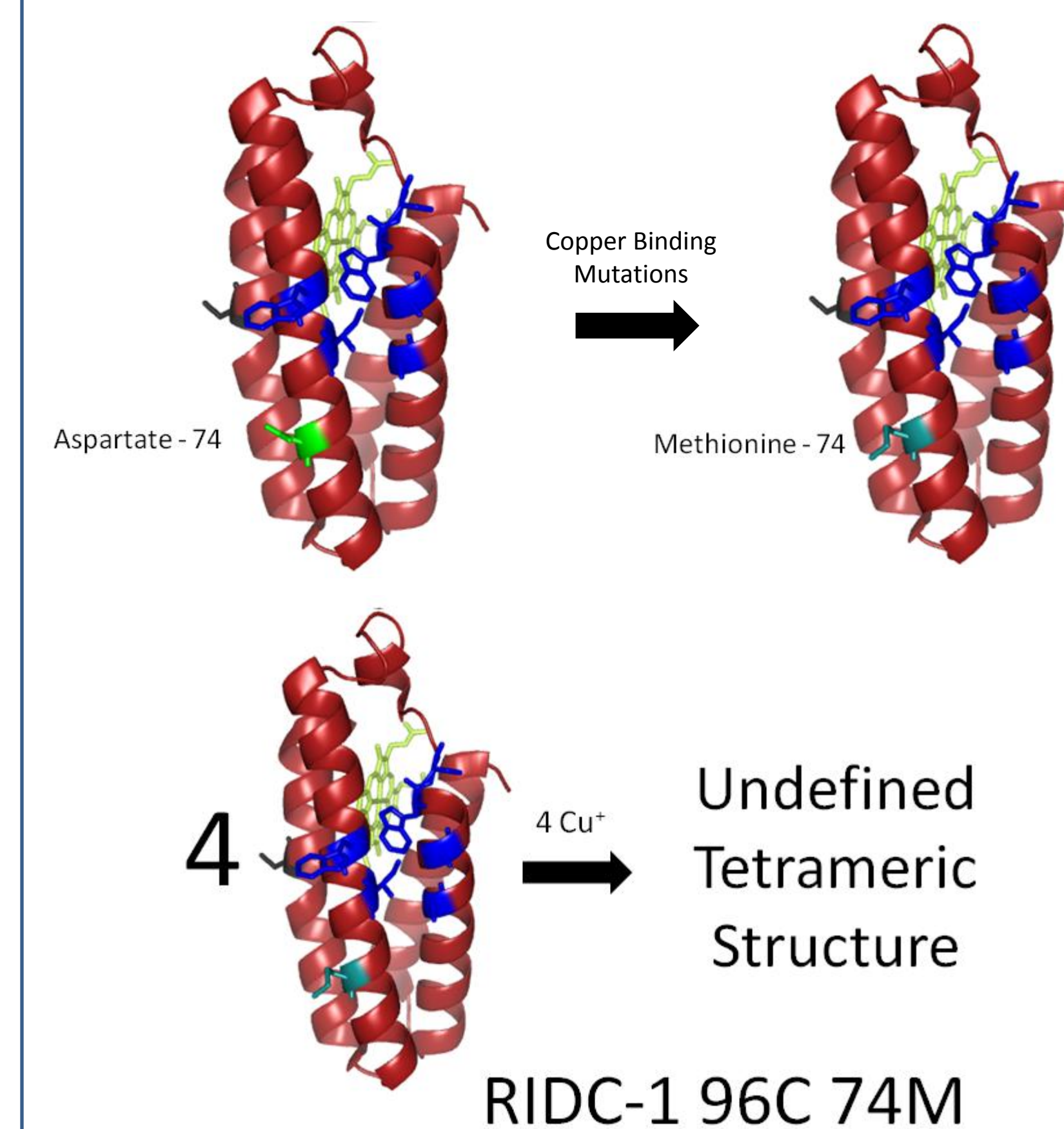
Interface Stabilization Based on Zn₄(MBPC-1)₄ Template (known as RIDC-1)



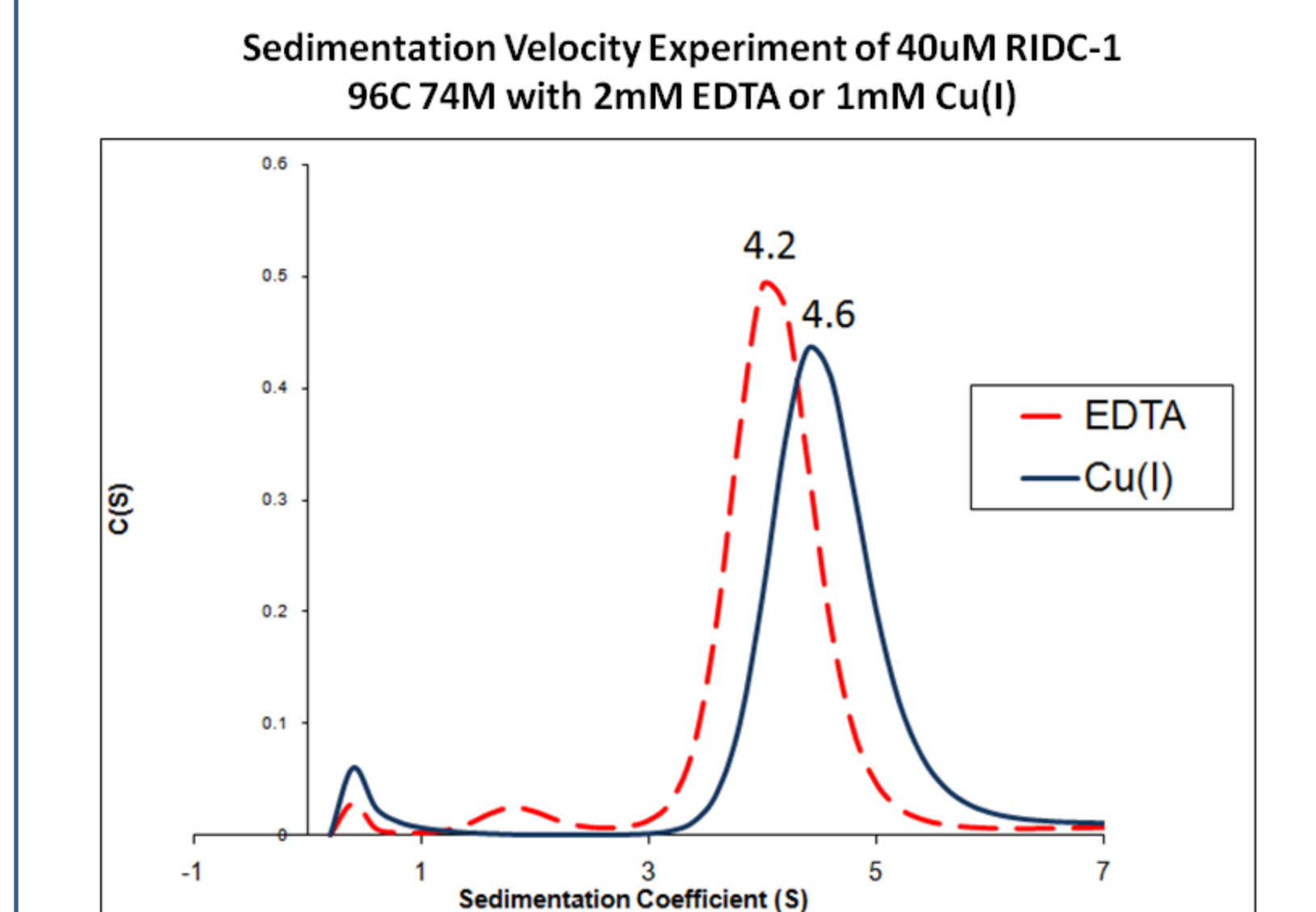
Covalent Linkage Between RIDC-1 Dimers (known as RIDC-1 96C)



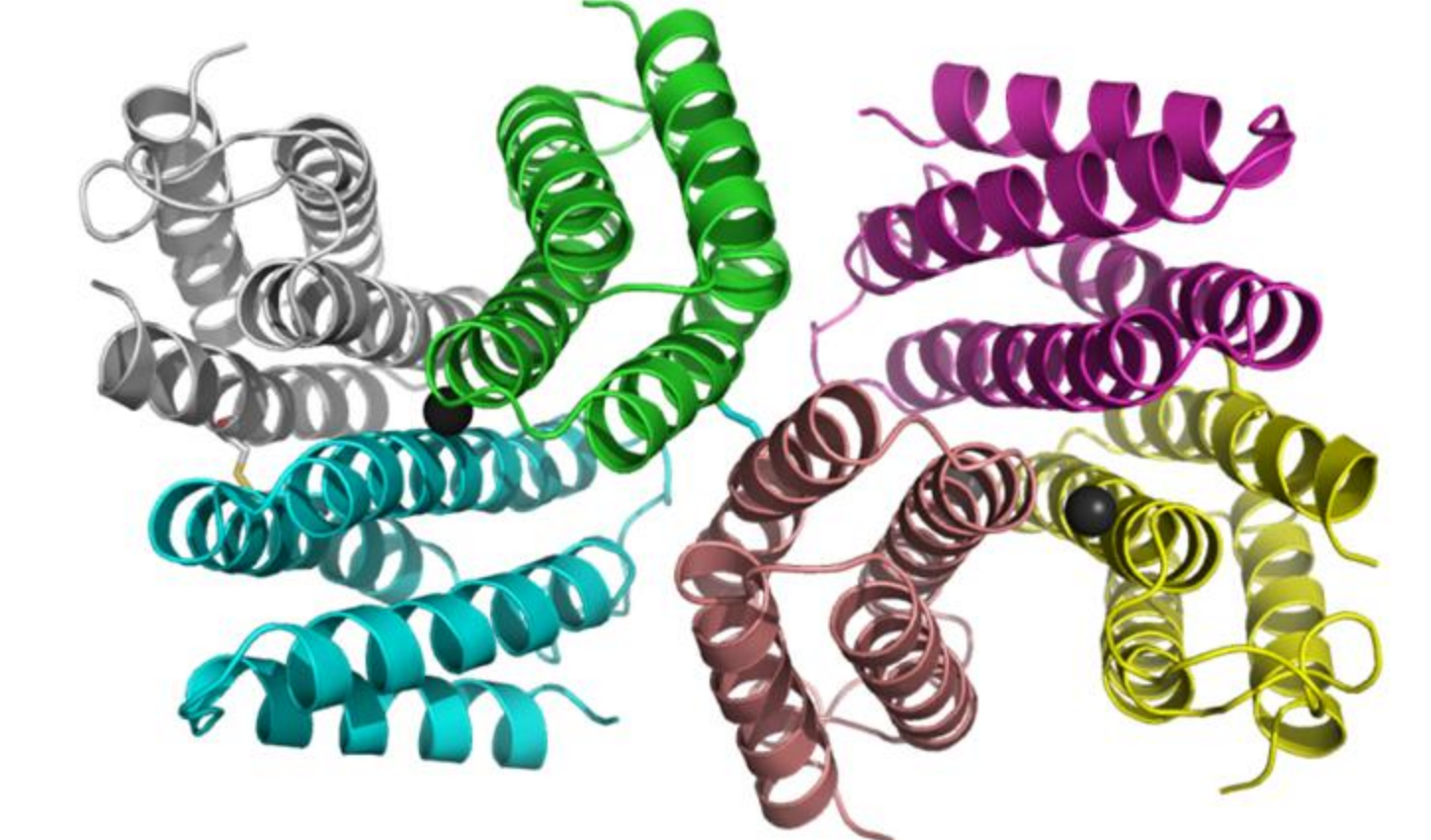
Introduction of Methionine for Cu(I) Binding (known as RIDC-1 96C 74M)



Experimentation Results on RIDC-1 96C 74M



Crystal Structure of RIDC-1 96C 74M in presence of Cu(I) (Determined with X-Ray Diffraction)



References

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2. KA Koch, MM Peña, DJ Thiele. *Chemistry and Biology*. 4(8) pp 549-560.
3. LH Hartwell, JJ Hopfield, S Liebler, AW Murray. 1999. From molecular to modular cell biology. *Nature* 402 (6761): C47-C52.
4. RJ Radford, PC Nguyen, TB Ditri, JS Figueroa, FA Tezcan. *Inorg. Chem.*, 49, 4362-4369

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